

5701 Cleveland St., Ste. 350 Virginia Beach, VA 23462 Tel 757.490.2242 • Fax 757.490.1548 • 800.486.1072

# Safety Data Sheet - Boric Acid

# **SECTION 1. Identification**

# 1.1. Product identifier

Boric Acid

 Index No
 005-007-00-2

 CAS
 10043-35-3

 ECN
 233-139-2

 REACH Registration number: 01-2119486683-25-0006

Trade names: Boric Acid

**Chemical name/synonyms:** Orthoboric acid, boric acid

1.2. Details of the supplier of the safety data sheet

Supplier Name: American Borate Company Address: 5701 Cleveland Street, Suite 350, Virginia Beach, VA 23462 Phone No: (757) 490-2242 or (800)-486-1072

#### **1.3. Emergency Phone Numbers:** Monday through Friday 8am – 5pm EST: (757) 490-2242 or (800) 486-1072 After 5pm and weekends: CHEMTREC 1-800-424-9300

**1.4. Recommended use of the chemical and restrictions on uses advised against**The product is used in industrial manufacturing, particularly in:CeramicsCosmeticsDetergentBorosilicate glassTextile fiberglass

There is no restriction on use of chemical.

# **SECTION 2. Hazard Identification**

### 2.1. Classification of the substance

**GHS Classification in accordance with 29 CFR 1910 (OSHA HCS)** Reproductive toxicity (Category 2), H361 Suspected of damaging fertility or the unborn child

2.2 GHS Label elements Pictogram Signal word Warning



### Hazard statements

H361 Suspected of damaging fertility or the unborn child.

### **Precautionary statements**

P201 Obtain special instructions before use.

**P202** Do not handle until all safety precautions have been read and understood.

**P281** Use personal protective equipment as required.

**P308/P313** If exposed or concerned: Get medical advice/ attention.

P405 Store locked up.

P501 Dispose of contents/ container to an approved waste disposal plant.

For Full text of R-S phrases as well as Hazard Class/Statements and Precautionary Statements see section 16.

### **Emergency overview**

Boric acid is a white odorless, powdered substance that is not flammable, combustible, or explosive, and has low acute oral and dermal toxicity.

### Potential health effects

Inhalation is the most significant route of exposure in occupational and other settings. Dermal exposure is not usually a concern because boric acid is poorly absorbed through intact skin.

### Inhalation

Occasional mild irritation effects to nose and throat may occur from inhalation of boric acid dusts at levels higher than 10 mg/m<sup>3</sup>.

### Eye contact

Boric acid is non-irritating to eyes in normal industrial use.

### Skin contact

Boric acid does not cause irritation to intact skin.

#### Ingestion

Products containing Boric Acid are not intended for ingestion. Boric Acid has a low acute toxicity. Small amounts (e.g., a teaspoon) swallowed accidentally are not likely to cause effects; swallowing amounts larger than that may cause gastrointestinal symptoms.

#### **Reproductive/developmental**

Animal ingestion studies in several species, at high doses, indicate that borates cause reproductive and developmental effects. A human study of occupational exposure to borate dust showed no adverse effect on reproduction. A recent epidemiological study and a peer reviewing report of the past epidemiological studies conducted in China didn't show any negative effect of boron on human fertility (10, 11).

### **Potential ecological effects**

Large amounts of Boric Acid can be harmful to plants and other species. Therefore, releases to the environment should be minimized.

### Signs and symptoms of exposure

Symptoms of accidental over-exposure to Boric Acid have been associated with ingestion or absorption through large areas of damaged skin. These may include nausea, vomiting and diarrhea, with delayed effects of skin redness and peeling.

Refer to section 11 for details on Toxicological data.

# **SECTION 3.** Composition/ Information on Ingredients

### 3.1. Substances:

Chemical Name	Purity	CAS	ECN	<b>REACH Registration No.</b>	Hazard Statement
Boric Acid	99.9%	10043-35-3	233-139-2	01-2119486683-25-0006	H361

### **SECTION 4. First-aid Measures**

### 4.1. Description of first aid measures

#### **General advice**

Move out of dangerous area. Seek medical attention. Show this safety data sheet to the doctor in attendance.

### Skin contact

Wash with soap and water. Seek medical attention.

### Eye contact

As with any chemical exposure to the eye, flush eyes with water for at least 20-minutes. Seek medical attention.

### Inhalation

If symptoms such as nose or throat irritation are observed, remove person to fresh air. If not breathing, give artificial respiration. Seek medical attention.

#### Ingestion

If large amounts are swallowed (i.e. more than one teaspoon), give two glasses of water or milk to drink and seek medical attention. Never give anything by mouth to an unconscious person.

### Note to physicians

Observation only is required for adult ingestion of less than 6 grams of boric acid. For ingestion in excess of 6 grams, maintain adequate kidney function and force fluids. Gastric lavage is recommended for symptomatic patients only. Hemodialysis should be reserved for massive acute ingestion or patients with renal failure. Boron analyses of urine or blood are only useful for documenting exposure and should not be used to evaluate severity of poisoning or to guide treatment (1) (see section 11).

### 4.2. Most important symptoms and effects, both acute and delayed

Described in labelling.

### 4.3. Indication of any immediate medical attention and special treatment needed.

No data available.

# **SECTION 5.** Fire-fighting Measures Identification

### 5.1. Suitable Extinguishing media

Use fire extinguishing media suitable for surrounding fires.

### 5.2. Specific hazards arising from the chemical

None - Boric acid is non-flammable, combustible or explosive. The product is itself a flame retardant.

### 5.3. Special protective actions for fire-fighters

Firefighters should wear pressure demand, self-contained breathing apparatus and full turn-out gear.

# **SECTION 6. Accidental Release Measures**

### 6.1. Personal precautions, protective equipment and emergency procedures

Avoid dust formation. Avoid breathing dust. In case of exposure to prolonged or high level of airborne dust, wear a personal respirator in compliance with national legislation.

### 6.2. Environmental precautions

Boric acid is a water-soluble white powder that may, at high concentrations cause damage to trees or vegetation by root absorption (see section 12). Do not flush to drains.

### 6.3. Methods and materials for containment and cleaning up

### Land spill

Vacuum, shovel or sweep up boric acid and place in containers for disposal in accordance with applicable local, state, and federal laws and regulations. Avoid contamination of water bodies during clean up and disposal. Avoid breathing dust.

### Spillage into water

Where possible, remove any intact containers from the water. Advise local water authority that none of the affected water should be used for irrigation or for the abstraction of potable water until natural dilution returns the boron value to its normal environmental background level (see sections 12, 13 and 15).

### 6.4. Reference to other sections

See sections 8 and 13 for further information.

# SECTION 7. Handling and Storage

### 7.1. Precautions for safe handling

To maintain package integrity and to minimize caking of the product, bags should be handled on a first-in first out basis. Good housekeeping and dust prevention procedures should be followed to minimize dust generation and accumulation. Use with appropriate local exhaust ventilation. The product should be kept away from strong reducing agents. Apply above handling advice when mixing with other substances.

### 7.2. Conditions for safe storage

Keep containers closed and store indoors in a dry well ventilated location. Provide appropriate ventilation and store bags such as to prevent any accidental damage.

### 7.3. Specific end use

See section 1.4.

### **SECTION 8. Exposure Controls/Personal Protection**

### 8.1. Control parameters

Occupational exposure limits for dust (total and resizable) are treated by OSHA, Cal OSHA and ACGIH as "Particulate Not Otherwise Classified" or "Nuisance Dust"

Respect regulatory provisions for dust (total and respirable).

ACGIH/TLV	10 mg/m <sup>3</sup>
Cal OSHA/PEL	10 mg/m <sup>3</sup>
OSHA/PEL (total dust)	15 mg/m <sup>3</sup>

### **DNEL values**

Exposure pattern	Type/site of effect	Exposure route	DNEL value
DNELs for workers			
Long-term	Systemic	Inhalation	8.3 mg BA/m <sup>3</sup>
Long-term	Systemic	Dermal	3924800 mg BA/day
DNELs for the general public			
Acute	Systemic	Oral	0.98 mg BA/kg bw/day
Long-term	Systemic	Dermal (external)	196 mg BA/kg bw/day
Long-term	Systemic	Dermal (systemic)	0.98 mg BA/kg bw/day
Long-term	Systemic	Inhalation	4.15 mg BA/m <sup>3</sup>
Long-term	Systemic	Oral	0.98 mg BA/kg bw/day

Source: Chemical Safety Report of Boric Acid

### PNEC values

PNEC add, freshwater, marine water = 1.35 mg B/L
PNEC add aqua intermittent = 9.1 mg B/L
PNEC add freshwater sediment, marine water sediment = 1.8 mg B/kg sediment dry weight
PNEC add, STP = 1.75 mg B/L
Source: Chemical Safety Report of Boric Acid

### 8.2. Exposure controls

### 8.2.1. Appropriate engineering controls

Maintain air concentrations below occupational exposure standards.

Use local exhaust ventilation to keep airborne concentrations of boric acid dust below permissible exposure levels. Wash hands before breaks and at the end of the workday. Remove and wash soiled clothing.

### 8.2.2. Individual protection measures, such as personal protective equipment

### Respiratory protection

Where risk assessment shows air-purifying respirators are appropriate use a full-face particle respirator type N100 (US) or type P3 (EN 143) respirator cartridges as a backup to engineering controls. If the respirator is the sole means of protection, use a full-face supplied air respirator. Use respirators and components tested and approved under appropriate government standards such as NIOSH (US) or CEN (EU).

*Eyes and hand protection* Handle with gloves. Wear eye protection suitable for job tasks.

### 8.2.3. Environmental exposure controls

No special requirement.

# **SECTION 9.** Physical and Chemical Properties

### 9.1. Information on basic physical and chemical properties

Physical state Color Odor Odor threshold Molecular weight Specific gravity powder or crystalline solid white odorless no data available 61.83 g/mole 1.51 gr/cm<sup>3</sup>

pH @ 20°C		6.1 (0.1 % solution) 5.1 (1.0% solution) 3.7 (4.7 % solution)
Melting point		$169 \pm 1^{\circ}C$ (heated in closed space)
Initial boiling point and boiling ra	ange	300 °C (full dehydration)
Flash point		Not applicable
Evaporation rate		Not applicable
Flammability (solid, gas)		Not flammable
Upper/lower flammability or exp	olosive limits	Not applicable
Vapor pressure		Negligible @ 20°C
Vapor density		No data available
Relative density		1.49 gr/cm <sup>3</sup> @ 23° C
Solubility in water		4.7% @ 20°C; 27.5% @ 100°C
Partition coefficient: n-octanol/v	vater	Log Kow (Pow): -1.09 at 22 °C
Auto-Ignition temperature		Not applicable
Decomposition temperature		No data available
Viscosity		No data available
Explosion hazard		Non explosive
Oxidizing properties		No data available
Bulk density:	Granular	49.94 lbs/ft <sup>3</sup> (0.80 ton/m <sup>3</sup> )
	Powder	37.46 lbs/ft <sup>3</sup> (0.60 ton/m <sup>3</sup> )

# **SECTION 10. Stability and Reactivity**

### 10.1. Reactivity

No data available

### 10.2. Chemical stability

Boric acid is stable under normal ambient and anticipated storage and handling conditions of temperature and pressure. When heated it loses water, first forming metaboric acid (HBO<sub>2</sub>), and on further heating it is converted into boric oxide  $(B_2O_3)$ .

### 10.3. Possibility of hazardous reactions

Reaction with strong reducing agents such as metal hydrides or alkali metals will generate flammable hydrogen gas which could create an explosive hazard.

### 10.4. Conditions to avoid:

Exposure to moisture and incompatible materials.

#### 10.5. Incompatible materials

Boric acid reacts as a weak acid which may cause corrosion of base metals. Avoid contact with strong reducing agents such as metal hydrides or alkali metals.

### 10.6. Hazardous decomposition products

Boranes, hydrogen, boron oxides.

# **SECTION 11.** Toxicological Information

### 11.1. Information on toxicological effect

### 11.1.1. Substances

### Acute toxicity

Low acute oral toxicity;  $LD_{50}$  in rats is 3,500 to 4,100 mg/kg of body weight.

### Skin corrosion / irritation

Low acute dermal toxicity; LD<sub>50</sub> in rabbits is greater than 2,000 mg/kg of body weight. Boric acid is poorly absorbed through intact skin. Non-irritating.

### Serious eye damage/ irritation

Non-irritating

Respiratory or skin sensitization:

Boric acid is not a skin sensitizer.

### Germ cell mutagenicity / carcinogenicity

IARC: No component of this product present at levels greater than or equal to 0.1% is identified as probable, possible or confirmed human carcinogen by IARC.

ACGIH: No component of this product present at levels greater than or equal to 0.1% is identified as a carcinogen or potential carcinogen by ACGIH.

NTP: No component of this product present at levels greater than or equal to 0.1% is identified as a known or anticipated carcinogen by NTP.

OSHA: No component of this product present at levels greater than or equal to 0.1% is identified as a carcinogen or potential carcinogen by OSHA.

### **Reproductive toxicity**

Animal feeding studies in rat, mouse and dog, at high doses, have demonstrated effects on fertility and testes (2). Studies in rat, mouse and rabbit, at high doses, demonstrate developmental effects on the fetus including fetal weight loss and minor skeletal variations. The doses administered were many times in excess of those which humans would normally be exposed to (3, 4, 5). Human epidemiological studies show no increase in pulmonary disease in occupational populations with chronic exposures to boric acid dust and sodium borate dust. A recent epidemiology study under the conditions of normal occupational exposure to borate dusts indicated no effect on fertility.

STOT-single exposure	N.A.
STOT-repeated exposure	N.A.

### Aspiration hazard

Low acute inhalation toxicity;  $LC_{50}$  in rats is greater than 2.0 mg/l (or g/m<sup>3</sup>).

### **SECTION 12. Ecological Information**

Boron occurs naturally in sea water at an average concentration of 5 mg B/I and fresh water at 1 mg B/I or less. In dilute aqueous solutions the predominant boron species present is undissociated boric acid. To convert boric acid into equivalent boron (B) content, multiply by 0.1748. Not persistent, not bio-accumulative.

### 12.1. Toxicity

### Phytotoxicity

Boron is an essential micronutrient for healthy growth of plants; however, it can be harmful to boron sensitive plants in higher quantities. Care should be taken to minimize the amount of borate product released to the environment.

### Algal toxicity (6)

Green algae, *Pseudokirchneriella subcapitata* (Hansveit and Oldersma, 2000) 72-hr EC50 –biomass = 40 mg B/L, or 229 mg boric acid/L.

### Invertebrate toxicity (7)

Daphnia, Daphnids, *Daphnia magna* (Gersich, 1984a) 48-hr LC50 = 133 mg B/L or 760 mg boric acid/L or 619 mg disodium tetraborate , anhydrous/L

### Fish toxicity (8)

Fish, Fathered minnow, *Pimephales promelas* (Soucek et al., 2010) 96-hr LC50 = 79.7 mg B/L or 456 mg boric acid/L or 370 mg disodium tetraborate, anhydrous

### 12.2. Persistence and degradability

Boron is naturally occurring and ubiquitous in the environment. Boric acid decomposes in the environment to natural borate.

### 12.3. Bio-accumulative potential

Not significantly bio-accumulative.

### 12.4. Mobility in soil

The product is soluble in water and is leachable through normal soil.

### 12.5. Results of PBT and vPvB assessment

12.6. Other adverse effects

# SECTION 13. Disposal Considerations

### 13.1. Disposal methods

Do not contaminate water, food, or feed by storage or disposal. Wastes resulting from the use of this product may be disposed of on site or at an approved waste disposal facility.

Non-refillable bag! Completely empty bag into application equipment. Offer for recycling, if available; otherwise dispose of empty bag in a sanitary landfill or by incineration, or if allowed by state and local authorities, by burning. If burning stay out of smoke.

### **SECTION 14. Transport Information**

Boric acid has no UN Number, and is not regulated under international rail, road, water or air transport regulations.

US DOT Not dangerous goods IMDG Not dangerous goods IATA Not dangerous goods

# SECTION 15. Regulatory Information

### 15.1. Safety, health and environmental regulations/substance specific legislation

It should be noted that borates are safe under conditions of normal handling and use, besides, they are essential nutrients to plants, and research shows that they play a beneficial role in human health.

### **Chemical inventory listing**

U.S. EPA TSCA Inventory	10043-35-3
Canadian DSL	10043-35-3
ECN	233-139-2
South Korea	1-439
Japanese MITI	(1)-63
Ensure all national/local reg	gulations are observed.

**SARA 302** No chemicals in this material are subject to the reporting requirements of SARA Title III, Section 302.

**SARA 313** This material does not contain any chemical components with known CAS numbers that exceed the threshold (De Minimis) reporting levels established by SARA Title III, Section 313.

# SARA 311/312 Hazards Chronic Health Hazard Massachusetts Right to Know Components

No components are subject to the Massachusetts Right to Know Act.

No data available

No data available

### Pennsylvania Right To Know Components

Boric acid CAS-No. 10043-35-3 Revision Date 2009-07-17

### New Jersey Right To Know Components

Boric acid CAS-No. 10043-35-3 Revision Date 2009-07-17

### California Prop. 65 Components

This product does not contain any chemicals known to State of California to cause cancer, birth defects, or any other reproductive harm.

### **Clean Air Act (Montreal Protocol)**

Boric acid was not manufactured with and does not contain any Class I or Class II ozone depleting substances.

### Cosmetics

The EC Directive 76/768/EEC sets an upper limit of 5% Boric acid in talc, 0.5% in oral hygiene products and 3% in other products. In addition, the talc should not be used on children under 3 years of age.

### **EU Reach Regulation**

Boric Acid is listed in the Candidate List of Substances of Very High Concern "SVHC" for eventual inclusion in Annex XIV to REACH Regulation 1907/2006 ("Authorization List"). (18.06.2010-ED/30/2010).

Boric acid is listed in the Annex XVII of REACH Regulation 1907/2006 (EU No.109/2012) and its use in consumer products above specific concentration limits is restricted. Note that this restriction is only specific to consumer products and do not cover its industrial and/or professional applications. Boric acid can be used in consumer products below specific concentration limits (which is C  $\geq$ 5.5% for Boric Acid).

### **15.2 FIFRA Statement**

This chemical is a pesticide product registered by the United States Environmental Protection Agency and is subject to certain labeling requirements under federal pesticide law. These requirements differ from the classification criteria and hazard information required for safety data sheets (SDS), and for workplace labels of non-pesticide chemicals. The hazard information required on the pesticide label is reproduced below. The pesticide label also includes other important information, including directions for use.

EPA Reg. No. 73605-2 EPA Est. 73605-VA-01

### **Precautionary Statements:**

**CAUTION**: Not for food or drug use **CAUTION**: Keep out of reach of children **CAUTION**: Harmful if swallowed.

**Hazards to Humans & Domestic Animals**: Causes moderate eye irritation. Avoid contact with eyes or clothing. Thoroughly was with soap and water after handling and before eating, drinking, chewing gum, using tobacco or using toilet.

**Environmental Hazards**: Boric Acid or boric acid solutions carelessly spilled or applied to cropland or growing plants, including trees and shrubs, may kill or seriously retard plant growth. Do not contaminate water by cleaning of equipment or disposal of wastes. Do not discharge effluent containing this product into lakes, streams, ponds, estuaries, oceans or other waters unless in accordance with the requirements of the National Pollutant Discharge Elimination System (NPDES) permit and the permitting authority has been notified in writing prior to discharge. Do not discharge effluent containing this product to sewer systems without previously notifying the local sewage treatment plant authority. For guidance, contact your State Water Board, or regional EPA office.

# **SECTION 16.** Other Information

### Full text of H-Statements referred to under sections 2 and 3.

H361 Suspected of damaging fertility or the unborn child

### References

1. Litovitz T L, Norman S A, Veltri J C, Annual Report of the American Association of Poison Control Centers Data Collection System. Am. J. Emerg. Med. (1986), 4, 427-458

2. Weir R J, Fisher R S, Toxicol. Appl. Pharmacol., (1972), 23, 351-364

3. National Toxicology Program (NTP) – Technical Report Series No. TR324, NIH Publication No. 88-2580 (1987), PB88 213475/XAB

- 4. Fail et al., Fund. Appl. Toxicol. (1991) 17, 225-239
- 5. Heindel et al., Fund. Appl. Toxicol. (1992) 18, 266-277
- 6. Hansveit and Oldersma, 2000; TNO Nutrition and Food Research Institute. Report No. V99.157.
- 7. Gersich, FM (1984a). Environ.Toxicol.Chem., 3 #1, 89-94 (1984)
- 8. Soucek et al., 2010. Illinois Natural History Survey, University of Illinois.
- 9. Birge W J, Black J A, EPA-560/-76-008 (April 1977) PB 267 085
- 10. Scialli AR, Bonde JP, Brüske-Hohlfeld I, Culver D, Li Y, Sullivan FM; ELSEVIER 2009
- 11. Robbins WA, Xun L, Jia J, Kennedy N, Elashoff DA, Ping L. ;ELSEVIER 2009;(Reproductive Toxicology)

For general information on the toxicology of borates see ECETOC Technical Report No. 63 (1995); Patty's Industrial Hygiene and Toxicology, 4th Edition Vol. II, (1994) Chap. 42, Boron.

### **Disclaimer of Liability**

The information in this SDS was obtained from sources which we believe are reliable. However, the information is provided without any warranty, express or implied, regarding its accuracy, reliability or completeness. The conditions or methods of handling, storage use or disposal of the product are beyond our control and may be beyond our knowledge. For this and other reasons, we do not assume responsibility and expressly disclaim liability for loss, damage or expense arising out of or in any way connected with the handling, storage, use or disposal of the product. It is the user's responsibility to satisfy themself as to the suitableness and completeness of such information for their own particular use. This SDS was prepared and is to be used only for this product. If the product is used as a component in another product, this SDS information may not be applicable. Please note that the provision of this SDS being not mandatory, only an English version of this latter is available.

Date of issue	8-2007
Date of Last Revision	6-2014